



Standard Test Method for Freezing Points of High-Purity Hydrocarbons¹

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1. Scope

1.1 This test method covers a procedure for the precise measurement of the freezing points of high-purity hydrocarbons.

1.2 The values stated in SI units are to be regarded as the standard. The values in parentheses are for information only.

1.3 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.* For specific hazard statements, see 5.1, 6.1 and 6.2.

NOTE 1—For the calculation of the molal purity of essentially pure compounds from measured freezing points and for procedures to be used for the sampling and determination of purity of certain specific compounds, see Test Method D1016.

2. Referenced Documents

2.1 ASTM Standards:²

D1016 Test Method for Purity of Hydrocarbons from Freezing Points

D1265 Practice for Sampling Liquefied Petroleum (LP) Gases, Manual Method

D4057 Practice for Manual Sampling of Petroleum and Petroleum Products

3. Summary of Test Method

3.1 The precise experimental measurement of the freezing point is made from interpretation of time-temperature freezing or melting curves.³

¹ This test method is under the jurisdiction of ASTM Committee D02 on Petroleum Products and Lubricants and is the direct responsibility of Subcommittee D02.04.0D on Physical and Chemical Methods.

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² For referenced ASTM standards, visit the ASTM website, www.astm.org, or contact ASTM Customer Service at service@astm.org. For *Annual Book of ASTM Standards* volume information, refer to the standard's Document Summary page on the ASTM website.

³ For details not given here, see Glasgow, A. R., Jr., Rossini, F. D., and Streiff, A. J., "Determination of the Purity of Hydrocarbons by Measurement of Freezing Points," *Journal of Research*, JNBAA, National Institute of Standards and Technology, Vol 35, No. 6, 1945, p. 355.

4. Significance and Use

4.1 The freezing point measured by this test method, when used in conjunction with the physical constants for the hydrocarbons listed in Test Method D1016, allows the determination of the purity of the material under test. A knowledge of the purity of these hydrocarbons is often needed to help control their manufacture and to determine their suitability for use as reagent chemicals or for conversion to other chemical intermediates or finished products.

5. Apparatus

5.1 *Freezing-Point Apparatus*,^{4,5} as shown in Figs. 1-3 comprising a freezing tube, a metal sheath for the freezing tube, a Dewar flask for the cooling bath, a Dewar flask for the warming bath, a stirring mechanism, suitable clamps and holders for the parts, and the absorption tubes. The outer walls of all Dewar flasks can be covered with adhesive tape to minimize danger from glass in case of breakage. (**Warning**—When using liquid nitrogen as a refrigerant, provide a means to prevent condensation of oxygen in the space between the freezing tube and the metal sheath and subsequent sealing of the space by ice forming on the ceramic (or glass) fiber collar. Provide the metal sheath with suitable openings in the *sides* and *bottom*. Failure to do this may result in breakage of the freezing tube when the liquefied oxygen evaporates within the sealed space.)

⁴ The sole source of supply of the apparatus known to the committee at this time is Reliance Glass Works, Inc., Bensenville, IL.

⁵ If you are aware of alternative suppliers, please provide this information to ASTM International Headquarters. Your comments will receive careful consideration at a meeting of the responsible technical committee,¹ which you may attend.