



Standard Test Method for Determination of Cooling Characteristics of Quenchants by Cooling Curve Analysis with Agitation (Drayton Unit)¹

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1. Scope

1.1 This test method covers the equipment and the procedure for evaluation of quenching characteristics of a quenching fluid by cooling rate determination.

1.2 This test method is designed to evaluate quenching fluids with agitation, using the Drayton Agitation Unit.

1.3 The values in SI units are to be regarded as the standard. The values in parentheses are for information only.

1.4 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.*

2. Referenced Documents

2.1 ASTM Standards:

E 220 Test Method for Calibration of Thermocouples by Comparison Techniques²

E 230 Temperature-Electromotive Force (EMF) Tables for Standardized Thermocouples²

2.2 SAE Standards:³

AMS 5665 Nickel Alloy Corrosion and Heat Resistant Bars, Forgings and Rings

2.3 Other Standards:⁴

Wolfson Engineering Group Specification Laboratory Tests for Assessing the Cooling Curve Characteristics of Industrial Quenching Media

3. Terminology

3.1 Definitions of Terms Specific to This Standard:

3.1.1 *aqueous polymer quenchant*—an aqueous polymer quenchant is an aqueous solution containing a water soluble polymer, typically including poly(alkylene glycol), poly(ethyl

oxazoline), poly(sodium acrylate), and poly(vinyl pyrrolidone) (1,2,3).⁵ The quenchant solution also typically contains additives for corrosion and foam control, if needed. Quench severity of aqueous polymer quenchants is dependent on concentration and molecular weight of the specific polymer being evaluated, quenchant temperature, and agitation rate as shown in Fig. 1, Fig. 2, and Fig. 3 respectively.

3.1.2 *cooling curve*—the cooling curve is a graphical representation of the cooling time (t) versus temperature (T) response of the probe (see 7.3). An example is illustrated in Fig. 4.

3.1.3 *cooling curve analysis*—the process of quantifying the cooling characteristics of a quenchant based on the temperature versus time profile obtained by cooling a preheated metal probe assembly (see Fig. 4) under standard conditions (1-7).

3.1.4 *cooling rate curve*—the cooling rate curve is a graphical representation of first derivative of the cooling curve, the rate of temperature change (dT/dt) versus temperature. An example is illustrated in Fig. 4.

3.1.5 *quenchant*—a quenching medium may be either a liquid or a gas. Gasses that are used as quenchants include air, nitrogen, argon, and hydrogen and, with the exception of air, which is used at atmospheric pressure, are used under pressure. Liquid quenchants include water, brine (most commonly dilute aqueous solutions of sodium chloride or sodium hydroxide), oil, molten salt, molten metal, and aqueous solutions of water soluble polymers. Water, brine, oil, and aqueous polymer quenchants are generally used with agitation.

3.1.6 *quench severity*—the ability of a quenching medium to extract heat from a hot metal (8).

4. Summary of Test Method

4.1 This test method determines the cooling time versus temperature of a standard nickel alloy probe assembly after it has been heated in a furnace to 850°C (1562°F) and then quenched in an aqueous polymer quenchant solution. The temperature inside the probe assembly and the cooling times are recorded at selected time intervals to establish a cooling temperature versus time curve. The resulting cooling curve (profile) may be used to evaluate quench severity (see Note 1).

¹ This test method is under the jurisdiction of ASTM Committee D02 on Petroleum Products and Lubricants and is the direct responsibility of Subcommittee D02.L0.06 on Nonlubricating Process Fluids.

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² *Annual Book of ASTM Standards*, Vol 14.03.

³ Available from Society of Automotive Engineers, 400 Commonwealth Dr., Warrendale, PA 15096.

⁴ Available from Wolfson Heat Treatment Centre, Aston University, Aston Triangle, Birmingham B4 7ET, England.

⁵ The boldface numbers in parentheses refer to the list of references at the end of this standard.