Designation: E 1786 – 02

Standard Test Method for Determination of Low Levels of Water in Liquid Chlorine by On-Line Infrared Spectrophotometry[1]

This standard is issued under the fixed designation E 1786; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon (ε) indicates an editorial change since the last revision or reapproval.

1. Scope

1.1 This test method is designed for the on-line determination of the content of water in liquid chlorine in the concentration range of 0.5 to 15 mg/kg (ppm).

1.2 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.* Specific hazards statements are given in Section 7 and Note 2.

2. Referenced Documents

2.1 *ASTM Standards:*

D 1193 Specifications for Reagent Water[2]

E 806 Test Method for Carbon Tetrachloride and Chloroform in Liquid Chlorine by Direct Injection (Gas Chromatography Procedure)[3]

2.2 *Federal Standards:*[4]

49 CFR 173 Code of Federal Regulations Title 49 Transportation: Shippers' General Requirements for Shipments and Packaging, including the following sections:

173.304 Charging of Cylinders with Liquefied Compressed Gas

173.314 Requirements for Compressed Gases in Tank Cars

173.315 Compressed Gases in Cargo Tanks and Portable Tank Containers

2.3 *Other Document:*[5]

Chlorine Institute Pamphlet No. 77—Sampling Liquid Chlorine

3. Summary of Test Method

3.1 Liquid chlorine continuously flows through a special infrared cell where it is maintained as a liquid under its own pressure. A process infrared spectrometer scans from 400 to 4400 wavenumbers of the infrared transmission spectrum of liquid chlorine. This spectrum then is ratioed to one obtained from the nitrogen-filled infrared cell previously. The ratioed spectrum is converted to absorbance, and the net absorbance of water band at 1596 wavenumbers, relative to a reference at 1663 wavenumbers, is determined.

3.2 The amount of water corresponding to this net absorbance is determined from a calibration curve prepared from the infrared absorbencies of standards which contain concentrations of water in liquid chlorine. These standards are prepared from manual samples of liquid chlorine in tantalum cylinders. Sample from each cylinder is introduced into a calibration infrared cell and maintained as a liquid under its own pressure.

4. Significance and Use

4.1 Trace amounts of water may be detrimental to the use of chlorine in some applications. The amount of water in the chlorine must be known to prevent problems during its use.

5. Apparatus

5.1 *Process Infrared Spectrometer*, capable of measurements in the 1600 wavenumber region. An FTIR with four wavenumber resolution is the instrument of choice, but dispersive instruments also may be used to achieve similar results.

5.2 *Special Infrared Calibration Cell* (Fig. 1), as used for calibration. Neither cell size nor pathlength are critical to the analysis, but sensitivity and limit of detection are dependent on pathlength. The concentration range reported in the Section 1 is achievable with a 60-mm pathlength cell constructed with the following: Figs. 2-7

5.2.1 *Hastelloy C and 316 Stainless Steel Stock*, suitable for machining,[6]

5.2.2 *Silver Chloride Windows*, 0.5 cm × 2.5 cm,[7] and

5.2.3 *Perfluoroelastomer Sheet*, 0.030 in thickness.[8]

5.3 *Ball Valves*, Monel[9] ¼-in. valve with pipe and ¼-in. tube ends.

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[1] This test method is under the jurisdiction of ASTM Committee E15 on Industrial and Specialty Chemicalsand is the direct responsibility of Subcommittee E15.02 on Product Standards.

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[2] *Annual Book of ASTM Standards*, Vol 11.01.

[3] *Annual Book of ASTM Standards*, Vol 15.05.

[4] *Code of Federal Regulations*, available from U.S. Government Printing Office, Washington, DC 20402.

[5] Available from The Chlorine Institute, Inc., 2001 L St. NW, Washington, DC 20036-4919.

[6] Available from Collins Instrument Company, Inc., P.O. Drawer 938, Angleton, TX 77516-0938.

[7] Available from Harshaw Chemical Company, 6801 Cochran Road, Solon, OH 44139.

[8] Available from E.I. DuPont de Nemours and Company, Wilmington, DE 19898.

[9] Available from International Nickel Company and has been found satisfactory for this purpose.

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